



ACHARYA INSTITUTE OF GRADUATE STUDIES

(NAAC Reaccredited 'A+' and Affiliated to Bengaluru City University)

Soladevanahalli, Bengaluru-560107

PROGRAM STRUCTURE FOR MASTER OF CHEMISTRY (MSc)

(With effect from the academic year 2025-26)

MSc – I SEMESTER

| Sl. No | Paper | Title of the paper | Instruction Hrs per Week | No. of Credits | Duration of the Exam. | Marks | | |
|--------|----------|-----------------------------------|--------------------------|----------------|-----------------------|---------------------|--------------------|-------------|
| | | | | | | Internal Assessment | Semester end exam. | Total Marks |
| 1 | Ch-101 | Inorganic chemistry-I | 4 | 4 | 3 Hrs | 30 | 70 | 100 |
| 2 | Ch-102 | Organic chemistry-I | 4 | 4 | 3 Hrs | 30 | 70 | 100 |
| 3 | Ch-103 | Physical Chemistry-I | 4 | 4 | 3 Hrs | 30 | 70 | 100 |
| 4 | Ch-104 | Analytical Chemistry | 4 | 4 | 3 Hrs | 30 | 70 | 100 |
| 5 | Ch-105 | Mathematics for Chemists | 3 | 2 | 3 Hrs | 30 | 70 | 100 |
| 6 | CP-106 | Inorganic chemistry practicals-I | 4 | 2 | 4 Hrs | 15 | 35 | 50 |
| 7 | CP-107 | Inorganic chemistry practicals-II | 4 | 2 | 4 Hrs | 15 | 35 | 50 |
| 8 | CP-108 | Physical chemistry practicals-I | 4 | 2 | 4 Hrs | 15 | 35 | 50 |
| 9 | CP - 109 | Physical chemistry practicals-II | 4 | 2 | 4 Hrs | 15 | 35 | 50 |
| | | Total | 35 | 26 | | | | 700 |



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MSc – II SEMESTER

| Sl. No | Paper | Title of the paper | Instruction Hrs per Week | No. of Credits | Duration of the Exam. | Marks | | |
|--------|----------|------------------------------------|--------------------------|----------------|-----------------------|---------------------|--------------------|-------------|
| | | | | | | Internal Assessment | Semester end exam. | Total Marks |
| 1 | Ch-201 | Inorganic chemistry-II | 4 | 4 | 3 Hrs | 30 | 70 | 100 |
| 2 | Ch-202 | Organic chemistry-II | 4 | 4 | 3 Hrs | 30 | 70 | 100 |
| 3 | Ch-203 | Physical Chemistry-II | 4 | 4 | 3 Hrs | 30 | 70 | 100 |
| 4 | Ch-204 | Molecular Spectroscopy | 4 | 4 | 3 Hrs | 30 | 70 | 100 |
| 5 | Ch-205 | Photochemistry | 3 | 2 | 3 Hrs | 30 | 70 | 100 |
| 6 | CP-206 | Inorganic chemistry practicals-III | 4 | 2 | 4 Hrs | 15 | 35 | 50 |
| 7 | CP-207 | Inorganic chemistry practicals-IV | 4 | 2 | 4 Hrs | 15 | 35 | 50 |
| 8 | CP-208 | Physical chemistry practicals-III | 4 | 2 | 4 Hrs | 15 | 35 | 50 |
| 9 | CP - 209 | Physical chemistry practicals-IV | 4 | 2 | 4 Hrs | 15 | 35 | 50 |
| | | Total | 35 | 26 | | | | 700 |



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MSc – III SEMESTER

| Sl. No | Paper | Title of the paper | Instruction Hrs per Week | No. of Credits | Duration of the Exam. | Marks | | |
|--------|---------|----------------------------------|--------------------------|----------------|-----------------------|---------------------|--------------------|-------------|
| | | | | | | Internal Assessment | Semester end exam. | Total Marks |
| 1 | Ch-301 | Organic Reaction Mechanisms | 4 | 4 | 3 Hrs | 30 | 70 | 100 |
| 2 | Ch-302 | Organic Synthesis | 4 | 4 | 3 Hrs | 30 | 70 | 100 |
| 3 | Ch-303 | Organic Spectroscopy | 4 | 4 | 3 Hrs | 30 | 70 | 100 |
| 4 | CP -305 | Organic chemistry practicals-I | 4 | 2 | 4 Hrs | 15 | 35 | 50 |
| 5 | CP -306 | Organic chemistry practicals-II | 4 | 2 | 4 Hrs | 15 | 35 | 50 |
| 6 | CP-307 | Organic chemistry practicals-III | 4 | 2 | 4 Hrs | 15 | 35 | 50 |
| 7 | CP-308 | Organic chemistry practicals-IV | 4 | 2 | 4 Hrs | 15 | 35 | 50 |
| | | Total | 28 | 20 | | | | 500 |



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MSc – IV SEMESTER

| Sl. No | Paper | Title of the paper | Instruction Hrs per Week | No. of Credits | Duration of the Exam. | Marks | | |
|--------|--------|---|--------------------------|----------------|-----------------------|---------------------|--------------------|-------------|
| | | | | | | Internal Assessment | Semester end exam. | Total Marks |
| 1 | Ch-401 | Stereochemistry and retrosynthetic analysis | 4 | 4 | 3 Hrs | 30 | 70 | 100 |
| 2 | Ch-402 | Chemistry of Natural Products | 4 | 4 | 3 Hrs | 30 | 70 | 100 |
| 3 | Ch-403 | Industrial Organic Chemistry | 4 | 4 | 3 Hrs | 30 | 70 | 100 |
| 4 | Ch-404 | Medicinal Organic Chemistry | 4 | 4 | 3 Hrs | 30 | 70 | 100 |
| 5 | CP-405 | Preparation of Industrially Important compounds | 4 | 2 | 4 Hrs | 15 | 35 | 50 |
| 6 | CP-406 | Extraction and Separations | 4 | 2 | 4 Hrs | 15 | 35 | 50 |
| 7 | CP-407 | Instrumental Methods and Quantitative Analysis | 4 | 2 | 4 Hrs | 15 | 35 | 50 |
| 8 | CP-408 | Qualitative Analysis of binary mixtures | 4 | 2 | 4 Hrs | 15 | 35 | 50 |
| | | Total | 32 | 24 | | | | 600 |



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NAME OF THE PROGRAM: **MASTER OF SCIENCE, CHEMISTRY**

COURSE OUTCOMES (CO'S)

M.Sc.: I Semester

SUBJECT NAME: CH 101: INORGANIC CHEMISTRY – I

After the completion of the Course, the students will be able to:

CO1: Explain and apply advanced concepts of chemical bonding including VSEPR theory, hybridization, electronegativity scales, lattice energy, and bonding models to predict molecular geometry and stability of inorganic compounds.

CO2: Analyze structures and bonding in main group compounds, boranes, carboranes, silicates, phosphazenes, and related cluster compounds using Wade's rules, MO theory, and electron-counting principles.

CO3: Evaluate acid–base behavior using HSAB theory, non-aqueous solvent systems, and polyoxometalates, and correlate these concepts with reactivity, stability, and applications.

CO4: Interpret electronic, magnetic, and nuclear phenomena including metal clusters, radioactive decay processes, and nuclear models, applying theoretical principles to inorganic and nuclear chemistry problems.

SUBJECT NAME: CH 102 - ORGANIC CHEMISTRY – I

After the completion of the Course, the students will be able to:

CO1: Illustrate bonding in organic molecules.

CO2: Explain the significance of reaction intermediates in organic chemistry.

CO3: Describe the importance of stereochemistry in organic compounds.

CO4: Understand the structure and function of carbohydrates and heterocyclic compounds.

SUBJECT NAME: CH 103 - PHYSICAL CHEMISTRY – I

After the completion of the Course, the students will be able to:

CO1: Apply quantum mechanics to explain atomic and molecular structures.

CO2: Interpret the principles of quantum mechanics and its models.

CO3: Analyze reaction kinetics, including theories of fast reactions.

CO4: Design kinetic models for homogeneous catalysis and surface chemistry.



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SUBJECT NAME: CH 104: ANALYTICAL CHEMISTRY – I

After the completion of the Course, the students will be able to:

CO1: Demonstrate knowledge of laboratory safety practices, chemical handling, error analysis, and statistical treatment of analytical data to ensure accuracy and precision in chemical measurements.

CO2: Apply classical quantitative analytical methods such as titrimetry and gravimetry to determine the composition of chemical samples with appropriate selection of indicators, reagents, and conditions.

CO3: Explain the principles of instrumental analytical techniques based on electromagnetic radiation, including Beer–Lambert’s law, spectrophotometric instrumentation, calibration methods, and detection limits.

CO4: Evaluate and apply separation techniques such as solvent extraction and chromatography (TLC, GC, HPLC) by analyzing factors affecting efficiency, resolution, and quantitative performance.

SUBJECT NAME: CH 105 - MATHEMATICAL CONCEPTS FOR CHEMISTS

After the completion of the Course, the students will be able to:

CO1: Solve linear equations and algebraic expressions.

CO2: Apply mathematical techniques in chemistry, including integrals for geometric and physical applications.

CO3: Solve differential equations in the context of chemical kinetics, assess the accuracy and precision of measurements



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M.Sc.: II Semester

SUBJECT NAME: CH 201: INORGANIC CHEMISTRY – II

After the completion of the Course, the students will be able to:

CO1: Explain metal–ligand equilibria, stability constants, and factors influencing complex formation using thermodynamic and kinetic principles and experimental determination methods.

CO2: Analyze coordination compounds using crystal field theory, molecular orbital theory, and angular overlap model to interpret geometry, electronic structure, colour, and magnetic properties.

CO3: Interpret electronic spectra of coordination complexes using Orgel and Tanabe–Sugano diagrams, and evaluate charge transfer processes and spectral properties of lanthanide and actinide complexes.

CO4: Assess magnetic and photochemical behavior of transition metal complexes, including spin crossover, exchange interactions, and photochemical reaction mechanisms relevant to inorganic systems.

SUBJECT NAME: CH 202 - ORGANIC CHEMISTRY – II

After the completion of the Course, the students will be able to:

CO1: Understand bonding in organic compounds.

CO2: Explain reaction mechanisms in substitution, addition, and elimination reactions.

CO3: Analyze rearrangement reactions and their mechanistic pathways.

CO4: Study vitamins and amino acids, their classification, and biological significance.

SUBJECT NAME: CH 203 - PHYSICAL CHEMISTRY – II

After the completion of the Course, the students will be able to:

CO1: Apply thermodynamic and statistical concepts to chemical systems.

CO2: Explain equilibrium and non-equilibrium thermodynamics.

CO3: Analyze electrochemical properties of solutions.

CO4: Apply electrochemistry concepts in real-world applications

SUBJECT NAME: CH 204: MOLECULAR SPECTROSCOPY – I

After the completion of the Course, the students will be able to:

CO1: Explain the interaction of electromagnetic radiation with matter and apply fundamental principles governing molecular energy levels and spectroscopic transitions.



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CO2: Interpret rotational and vibrational spectra of molecules to determine molecular structure, force constants, and bonding characteristics.

CO3: Analyze electronic spectra and apply selection rules, transition probabilities, and spectral parameters to molecular and coordination compounds.

CO4: Correlate spectroscopic data from different techniques to solve structural and analytical problems in chemistry with scientific accuracy.

SUBJECT NAME: CH 205 - PHOTOCHEMISTRY

After the completion of the Course, the students will be able to:

CO1: Understand fundamental photochemical principles, laws, and excited-state chemistry.

CO2: Analyze photochemical reactions and emission spectra.

CO3: Study photosensitized reactions, photodegradation, and photovoltaic effects.



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M.Sc.: III Semester

SUBJECT NAME: CH 301 - ORGANIC REACTION MECHANISMS

After the completion of the Course, the students will be able to:

- CO1:** Explain the fundamental principles of aliphatic substitution reactions, explain the mechanism of its.
- CO2:** Describe reactivity and mechanism of photochemical reactions
- CO3:** Analyse molecular orbital symmetry, Frontier molecular approach concepts and discuss the pericyclic reactions.
- CO4:** Discuss the free radicals' reactions, and its mechanisms and explain the role of coenzymes in the in biochemical mechanism

SUBJECT NAME: CH 302 - ORGANIC SYNTHESIS

After the completion of the Course, the students will be able to:

- CO1:** To acquire the knowledge of named organic reactions in C-C and C-N bond-forming reactions.
- CO2:** Describe the preparation and applications of the organic reagents in organic synthesis and Functional group transformation.
- CO3:** Discuss oxidizing agents and reducing agents in organic synthesis.
- CO4:** Utilize the principles of enantioselectivity and diastereoselective in asymmetric synthesis.

SUBJECT NAME: CH 303 - ORGANIC SPECTROSCOPY

After the completion of the Course, the students will be able to:

- CO1:** Enumerate the theory, principle and applications of Ultraviolet-visible and Vibrational (IR) spectroscopy.
- CO2:** Understand the principle and applications of Nuclear magnetic resonance spectroscopy.
- CO3:** Understand the principle of mass spectrometry and its applications.
- CO4:** Interpret the spectroscopic data for structure determination of unknown organic compounds.



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M.Sc.: IV Semester

SUBJECT NAME: CH 401 - STEREOCHEMISTRY AND RETROSYNTHETIC ANALYSIS

After the completion of the Course, the students will be able to:

CO1: Explain the fundamental principles of aliphatic substitution reactions, explain the mechanism of its.

CO2: Describe reactivity and mechanism of photochemical reactions

CO3: Analyse molecular orbital symmetry, Frontier molecular approach concepts and discuss the pericyclic reactions.

CO4: Discuss the free radicals' reactions, and its mechanisms and explain the role of coenzymes in the in biochemical mechanism

SUBJECT NAME: CH 402 - CHEMISTRY OF NATURAL PRODUCTS

After the completion of the Course, the students will be able to:

CO1: Discuss the classes, methods of isolation, stereochemistry, structural elucidation of terpenoids, and Sketch the synthesis and biosynthesis of terpenoids

CO2: Identify the classes, discuss the methods of isolation, stereochemistry, and structural elucidation of Alkaloids and sketch the synthesis and biosynthesis of alkaloids

CO3: Elucidate the structure of Porphyrins and Nucleic acids and sketch the synthesis of its

CO4: Explain the classes, biological role, stereochemistry, structural elucidation of prostaglandins and sketch the synthesis, biosynthesis of prostaglandins and insect pheromones

SUBJECT NAME: CH 403 - INDUSTRIAL ORGANIC CHEMISTRY

After the completion of the Course, the students will be able to:

CO1: Describe the methods of applying dye to fabric and Sketch the synthesis and application of dyes

CO2: Discuss the classification, chemical properties, synthesis and mechanism of heterocyclic and mesoionic compounds

CO3: Recognize the typical organometallic reagent, explain their utility in organic synthesis and interpret its mechanism

CO4: Discuss the nomenclature, properties, stereochemistry, techniques of polymerization. Processing techniques, spinning and sketch the synthesis and applications of polymers.



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SUBJECT NAME: CH 404 - MEDICINAL ORGANIC CHEMISTRY

After the completion of the Course, the students will be able to:

CO1: Use QSAR and computational techniques in drug discovery.

CO2: Analyze medicinally important organic compounds.

CO3: Apply synthetic strategies in pharmaceutical chemistry.

CO4: Design commercial synthetic routes for drug molecules.
